

8942 version 3
Characterise the nature of chemical systems at equilibrium
Level 2 Credits 2

Characterise the nature of chemical systems at equilibrium

- Equilibrium constant expressions, K (or K_c) are written for *given* chemical equations.
- Changes in equilibrium position are predicted as a result of a change in conditions.
 - Effect on amount of named reactant / product
 - Effect on the equilibrium position
 - Change of temperature
 - Change of concentration
 - Change of pressure
 - Addition / removal of catalyst.

Notes

